

SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR
(AUTONOMOUS)

B.Tech I Year I Semester Supplementary Examinations February-2024

BASIC THERMODYNAMICS

(Agricultural Engineering)

Time: 3 Hours

Max. Marks: 60

(Answer all Five Units 5 x 12 = 60 Marks)

UNIT-I

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|---|--|-----|-----------|----|
| 1 | a What is a thermodynamic system? Explain different types of systems with suitable examples. | CO1 | L1&
L2 | 6M |
| | b What do mean by property? Distinguish between intensive and extensive property. | CO1 | L1 | 6M |

OR

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|---|---|-----|----|----|
| 2 | a Compare work transfer and heat transfer with neat sketches. | CO1 | L5 | 6M |
| | b Show that work is a path function and not a property. | CO1 | L1 | 6M |

UNIT-II

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| 3 | What is Steady Flow Process? Derive Steady Flow Energy Equation(SFEE) for an open system. | CO2 | L1&
L3 | 12M |
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OR

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| 4 | a One kg of Air is heated from 200C to 1050 C. Find the change of internal energy and change of enthalpy. Assume $C_p=1.01$ KJ/KgK and $C_v=0.72$ KJ/KgK. | CO2 | L3 | 6M |
| | b In an air motor cylinder the compressed air has an internal energy of 450 kJ/kg at the beginning of the expansion and an internal energy of 220 kJ/kg after expansion. If the work done by the air during the expansion is 120 kJ/kg, calculate the heat flow to and from the cylinder. | CO2 | L4 | 6M |

UNIT-III

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| 5 | Derive the relation for Work done, Heat Transfer and Change enthalpy of an Isochoric Process. | CO3 | L3 | 12M |
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| 6 | A piston-cylinder arrangement contains 0.05 m ³ of nitrogen at 1 bar and 280 K. The piston moves inwards and the gas is compressed isothermally and reversibly until the pressure becomes 5 bar. Determine :(i) Change in entropy. (ii) Workdone. Assume nitrogen to be a perfect gas. | CO3 | L3 | 12M |
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UNIT-IV

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| 7 | Develop the expression for air standard efficiency for diesel engine. | CO4 | L6 | 12M |
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| 8 | Build the phase equilibrium diagram for a pure substance P-V , P-T, T-S plot with relevant constant property line. | CO4 | L3 | 12M |
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UNIT-V

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| 9 | Describe the different operations of Rankine cycle and also derive the expression for its efficiency . | CO5 | L1 | 12M |
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| 10 | Explain the process of improving Rankine cycle efficiency with regeneration. | CO5 | L2 | 12M |
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